Theor Chim Acta (1992) 83: 5-13

Theoretica Chimica Acta

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Introduction of Klaus Ruedenberg*

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Received June 5, 1991/Accepted August 26, 1991

I am honored to have been asked to introduce Klaus Ruedenberg, a longtime companion in the search.

The first time I introduced Klaus was thirty-seven or thirty-eight years ago, when he was about to deliver a seminar on exchange integrals to my research group beginning at eleven one Monday morning. Fortunately I had had a hunch and had warned him: I should have to attend a committee meeting at two in the afternoon. At two o'clock Klaus was still going on, having covered four blackboards with formulas; I quietly slipped out the door. Tonight we may have the same problem: I have a commitment at 8:30 sharp tomorrow morning. Klaus, please leave me a little time to sleep!

Briefly, then, who is our speaker this evening?

Born in Bielefeld, Germany in 1920, Klaus Ruedenberg received his early education there. In 1938, Klaus was able to move to Switzerland and thereby escape the persecution to which his father fell victim. Until 1948 he resided in Switzerland, studying first chemistry and then physics. He ultimately received his Ph.D in physics from the University of Zürich in 1950, after he had already moved to the University of Chicago with his thesis supervisor Gregor Wentzel. He worked fruitfully in the laboratory of Robert Mulliken from 1950 until 1955, when he joined Iowa State University as Assistant Professor of Chemistry. Except for two years elsewhere he has been here ever since, becoming Distinguished Professor in 1978. In 1975 Klaus received an honorary degree from the University of Basel; this very year he will be receiving another honorary degree from the University of Bielefeld. He has received the Midwest Award of the American Chemical Society and is a member of the International Academy of Quantum Molecular Science. He is, of course, the valiant Editor-in-Chief of Theoretica Chimica Acta.

Up to the present moment, Klaus has worked with thirty graduate students and postdoctorals, about half of whom are participating in this symposium. He

^{*} Introduction of Klaus Ruedenberg on the occasion of his lecture at the International Symposium on *Ab Initio* Methods in Quantum Chemistry, Ames, Iowa, May 10, 1991

and his associates have published more than one hundred and fifty scientific papers; these are listed below. These contributions are penetrating and varied, clearly in total constituting a major accomplishment – some of the best theoretical chemistry of our time.

In a few minutes it is impossible to adequately describe Klaus Ruedenberg's achievements. I will venture just a few words, about just a few of his papers.

Klaus' first publication was his doctoral research with Gregor Wentzel. You see what his background was when, in 1950 at the age of thirty, he joined Robert Mulliken, John Platt, and Clemens Roothaan in that great LMSS at the University of Chicago. These four persons influenced Klaus greatly, I believe, particularly John Platt, somebody whom the chemical physics community has much missed since he left us for the career of a futurist.

The second publication of Klaus Ruedenberg, his first in quantum chemistry, was in fact coauthored with me. [Forgive me, Klaus, I hope that this does not embarrass you.] This paper resulted from a month that he spent at Carnegie Tech in the fall of 1950. His technical provess and physical soundness were immediately evident to me, and I have been his admirer through the decades since then.

The third paper by Klaus is the first of his wonderful series on molecular integrals. Some of you may not realize what a feat this and the subsequent papers constitute. Try, yourself, just to evaluate the exchange integral between two equivalent 1s orbitals a distance R apart. Heitler couldn't do it; London couldn't do it. Suguira did it, but to retrace his steps requires a week of work. Introducing powerful new techniques, Ruedenberg did it and went far beyond, to nonequivalent s orbitals, p and d orbitals, and (subsequently) to three- and four-center integrals.

This level of difficulty, this level of success; that is what separates the men from the boys. No wonder Klaus spoke on the subject for three hours as I have already described! And somebody at this symposium yesterday had the gall to say, in effect, that modern quantum chemistry began in the 1970's!

The fourth paper I remark on is Klaus' Publication 8. This is a *tour de force* in which Klaus succeeded in establishing an isomorphism between the free-electron network model for conjugated *pi*-electron systems, including a beautiful representation of electron repulsion within that model, with the LCAO theory of the so-called PPP form. The term "zero differential overlap" was coined by Klaus, incidentally.

Klaus' 1962 Publication 20 and his 1963 Publication 23 are famous, the first for carefully expounding the interplay between kinetic and potential energies during chemical bond formation, the second for establishing the viability of defining localized orbitals as those Hartree–Fock orbitals that maximize the self-coulomb repulsion or minimize the interorbital exchange.

It was just at this point in his career, in 1962, that Klaus for a short time took up a professorship at Johns Hopkins University. [These were days when the quality of one's publications counted for more than their number, it may be noted in passing.] But he preferred Iowa State, most particularly the advantages of being on a faculty that included Robert Rundle, and he quickly returned to Ames.

In my opinion six of Klaus' papers over the next twenty years are particularly important. These are his Publication 33 (electron pair wavefunction for Be), Publication 46 (vibrational spectroscopic constants), Publication A22 (eventempered orbitals), Publication 83 (relation between total energy and sum of

orbital energies), Publication 87 (MCSCF optimization), and Publication 98 (electron difference densities). Each of these papers is definitive, each is breaking into new territory, and each has led to many later developments. As for Klaus Ruedenberg's contributions from 1985 on, most of you here at this conference are intimately familiar with his succession of fine contributions to the understanding of potential energy surfaces, the latest of which he will be telling us about.

Ladies and gentlemen, here then is a consummate professional; here is a world leader in quantum chemistry. I give you Klaus Ruedenberg. In his person will be found soft-spoken confidence, staying power, guts, zest; that is to say, a certain class. In his writing will be found clarity, thoroughness, elegance; that is to say, a certain style. In his works will be found depth, great scope; jewels; that is to say, bits sculpted for the years. I present to you Klaus Ruedenberg: exemplary physicist, exemplary chemist, supreme expositor, teacher, and friend. One of the men, not one of the boys.

Publications of Klaus Ruedenberg 1951–1991

- 1. On the theory of strong coupling between nucleons and pseudovector mesons. Ruedenberg K (1951) Helv Phys Acta 24:89
- 2. A mobile electron model for aromatic molecules. Ruedenberg K, Parr RG (1951) J Chem Phys 19:1268
- 3. A study of the two-center exchange integrals in molecular problems. Ruedenberg K (1951) J Chem Phys 19:1459
- 4. A note on the three- and four-center integrals in molecular problems. Ruedenberg K (1951) J Chem Phys 19:1433
- 5. Free-electron network model for conjugated systems. I. Theory. Ruedenberg K, Scherr CW (1953) J Chem Phys 21:1565
- Free-electron network model for conjugated systems. V. Theoretical equivalence with the LCAO MO model. Ruedenberg K (1954) J Chem Phys 22:1878
- 7. A study of the two-center hybrid integrals and a unified treatment of the hybrid, Coulomb, and one-electron integrals. Ruedenberg K, Roothaan CCJ, Jaunzemis W (1956) J Chem Phys 24:201
- 8. Electronic interaction in the free-electron network model for conjugated systems. I. Theory, Ham NS, Ruedenberg K (1956) J Chem Phys 25:1
- 9. Electronic interaction in the free-electron network model for conjugated systems. II. Spectra of aromatic hydrocarbons. Ham NS, Ruedenberg K (1956) J Chem Phys 25:13
- Energy levels, atom populations, bond populations in the LCAO MO model and in the FE MO model. A quantitative analysis. Ham NS, Ruedenberg K (1958) J Chem Phys 29:1199
- 11. Mobile bond orders in conjugated systems. Ham NS, Ruedenberg K (1958) J Chem Phys 29:1215
- 12. Theorem on the mobile bond orders of alternant conjugated systems. Ruedenberg K (1958) J Chem Phys 29:1232
- 13. Boulder Conference on Molecular Quantum Mechanics. Ruedenberg K (1960) Physics Today 13(5):34
- 14. Quantum mechanics of mobile electrons in conjugated bond systems. I. General analysis of the tight-binding approximation. Ruedenberg K (1961) J Chem Phys 34:1861
- 15. Quantum mechanics of mobile electrons in conjugated bond systems. II. Augmented tight-binding approximation. Ruedenberg K (1961) J Chem Phys 34:1878
- 16. Quantum mechanics of mobile electrons in conjugated bond systems. III. Topological matrix as generatrix of bond orders. Ruedenberg K (1961) J Chem Phys 34:1884
- 17. Quantum mechanics of mobile electrons in conjugated bond systems. IV. Integral formulas. Ruedenberg K (1961) J Chem Phys 34:1892
- Quantum mechanics of mobile electrons in conjugated bond systems. V. Empirical determination of integrals between carbon atomic orbitals from experimental data on benzene. Ruedenberg K, Layton EM Jr (1961) J Chem Phys 34:1897

- 19. Quantum mechanics of mobile electrons in conjugated bond systems. VI. Theoretical evaluation of energy contributions. Ruedenberg K (1961) J Chem Phys 34:1907
- 20. The physical nature of the chemical bond. An interpretation of cohesion in polyatomic many-electron systems. Ruedenberg K (1962) Revs Modern Phys 34:326
- Ligand field theory of square-planar platinum(II) complexes. Fenske RF, Martin DS, Ruedenberg K (1962) Inorg Chem 1:441-452
- 22. Electronic structure and spectra of conjugated hydrocarbons. Hummel RL, Ruedenberg K (1962) J Phys Chem 66:2334
- Localized atomic and molecular orbitals. Edmiston C, Ruedenberg K (1963) Revs Modern Phys 35:457
- 24. Two-center electron interaction energies. Ruedenberg K (1964) in: Löwdin PO, Pullman B (eds) Molecular orbitals in chemistry, physics and biology. Academic Press, NY, p 215
- Hindered rotation, Hellmann-Feynman theorem, and localized molecular orbitals. Ruedenberg K (1964) J Chem Phys 41:588
- 26. Chemical binding in the water molecule. Edmiston C, Ruedenberg K (1964) J Phys Chem 68:1628
- 27. Chemical binding in diatomic hydrides. Layton EM Jr, Ruedenberg K (1964) J Phys Chem 68:1654
- 28. Chemical binding in homonuclear diatomics. Rue RR, Ruedenberg K (1964) J Phys Chem 68:1676
- 29. Free-electron network for conjugated systems. VII. A note on the joint condition. Ruedenberg K (1964) in: Free-electron theory of conjugated molecules. Wiley, NY, Sect 8
- Three-dimensional and one-dimensional free-electron molecular orbitals. Hummel RL, Ruedenberg K (1965) in: Sinanoglu O (ed) Istanbul Lectures on Quantum Chemistry. Academic Press, NY, Vol 1, p 113
- Localized self-consistent-field orbitals in atoms and molecules. Ruedenberg K (1965) in: Sinanoglu O (ed) Istanbul Lectures on Quantum Chemistry. Academic Press, NY, Vol 1, p 85
- 32. Localized atomic and molecular orbitals. II. Edmiston C, Ruedenberg K (1965) J Chem Phys 43:S97
- Electron correlation and electron pair wavefunction for the beryllium atom. Miller K, Ruedenberg K (1965) J Chem Phys 43:S88
- 34. Overlap integrals between atomic orbitals. Ruedenberg K, Oohata K, Wilson D (1966) J Math Phys 7:539
- 35. Two-center coulomb integrals between atomic orbitals. Oohata K, Ruedenberg K (1966) J Math Phys 7:547
- 36. Localized atomic and molecular orbitals. III. Edmiston C, Ruedenberg K (1966) in: Löwdin PO (ed) Quantum theory of atoms, molecules and the solid state. Academic Press, NY, p 263
- Bipolare Entwicklungen, Fourier Transformation und molekulare Mehrzentren-Integrale. Ruedenberg K (1967) Theor Chim Acta 7:359
- Two-center hybrid integrals between Slater-type atomic orbitals. Christoffersen RE, Ruedenberg K (1967) J Chem Phys 47:1855
- 39. Electron correlation and separated pair approximation in beryllium-like atomic systems. Miller KJ, Ruedenberg K (1968) J Chem Phys 48:3414
- 40. Electron correlation and the augmented separated pair approximation. Miller KJ, Ruedenberg K (1968) J Chem Phys 48:3444
- Electron correlation and augmented separated pair approximation in beryllium-like atomic systems. Miller KJ, Ruedenberg K (1968) J Chem Phys 48:3450
- 42. Analysis and evaluation of two-center hybrid integrals over Slater-type atomic orbitals. Christoffersen RE, Ruedenberg K (1968) J Chem Phys 49:4285
- 43. Atomic orbital overlap integrals. Silver DM, Ruedenberg K (1968) J Chem Phys 49:4301
- Coulomb integrals between Slater-type atomic orbitals. Silver DM, Ruedenberg K (1968) J Chem Phys 49:4306
- 45. A new form of the bipolar expansion and molecular multicenter integrals. Salmon LS, Birss FW, Ruedenberg K (1968) J Chem Phys 49:4293
- Accurate correlations between rotational and vibrational spectroscopic constants in diatomic molecules. Calder GV, Ruedenberg K (1968) J Chem Phys 49:5399

- 47. Two-center exchange integrals between Slater-type atomic orbitals. Mehler EL, Ruedenberg K (1969) J Chem Phys 50:2575
- Compact natural orbital expansions of the helium ground state. Cressy N, Miller KJ, Ruedenberg K (1969) Int J Quantum Chem 3:107
- Expansion of r₁₂ and r₁₂⁻¹ in terms of analytical functions. Cressy N, Ruedenberg K (1969) Int J Quantum Chem 3:493
- The origin of binding and antibinding in the hydrogen molecule ion. Feinberg MJ, Ruedenberg K, Mehler EL (1970) in: Löwdin PO (ed) Advances in Quantum Chemistry, Vol 5. Academic Press, NY, p 27
- Electron correlation and separated pair approximation in diatomic molecules. I. Theory. Silver DM, Mehler EL, Ruedenberg K (1970) J Chem Phys 52:1174
- 52. Electron correlation and separated pair approximation in diatomic molecules. II. Lithium hydride and boron hydride. Mehler EL, Ruedenberg K, Silver DM (1970) J Chem Phys 52:1181
- Electron correlation and separated pair approximation in diatomic molecules. III. Imidogen. Silver DM, Ruedenberg K, Mehler EL (1970) J Chem Phys 52:1206
- 54. Parametrization of an orthogonal matrix. Raffenetti R, Ruedenberg K (1970) Int J Quantum Chem 3S:625
- 55. Paradoxical role of the kinetic energy operator in the formation of the covalent bond. Feinberg MJ, Ruedenberg K (1971) J Chem Phys 54:1495
- 56. Translation of Slater-type atomic orbitals. England W, Ruedenberg K (1971) J Chem Phys 54:2291
- 57. Numerical analysis and evaluation of normalized repeated integrals of the error function and related functions. Bardo RD, Ruedenberg K (1971) J Comput Phys 8:167
- Localized molecular orbitals: A bridge between chemical intuition and molecular quantum mechanics. England W, Salmon LS, Ruedenberg K (1971) Fortschritte der Chemischen Forschung 23:31
- 59. Localized pi-orbitals, Pauling bond orders and the origin of aromatic stability. England W, Ruedenberg K (1971) Theoret Chim Acta 22:196
- 60. Heteropolar one-electron bond. Feinberg MJ, Ruedenberg K (1971) J Chem Phys 55:5804
- 61. Expectation values of many-fermion spin eigenstates. Ruedenberg K (1971) Phys Rev Lett 27:1105
- 62. Quadrupolar expansion for r_{12}^{-1} . Salmon LS, Ruedenberg K (1972) Int J Quantum Chem 6:347
- An expansion for four-center integrals over Slater-type atomic orbitals. Salmon LS, Ruedenberg K (1972) Int J Quantum Chem 6:353
- 64. Generalization of Euler angles to N-dimensional orthogonal matrices. Hoffman DK, Raffenetti RC, Ruedenberg K (1972) J Math Phys 13:528
- 65. Matrix elements and density matrices for many-electron spin eigenstates built from orthonormal orbitals. Ruedenberg K, Poshusta RD (1972) in: Löwdin PO (ed) Advances in Quantum Chemistry, Vol 6. Academic Press, NY, p 267
- 66. Molecular one-electron integrals over Slater-type atomic orbitals and irregular solid spherical harmonics. Steinborn EO, Ruedenberg K (1972) Int J Quantum Chem 6:413
- 67. Many-electron wavefunctions expanded in spin-adapted antisymmetrized products and their expectation values. Salmon WI, Ruedenberg K (1972) J Chem Phys 57:2776
- Implementing the SAAP formalism I. Serber-type spin eigenfunctions by direct diagonalization. Salmon WI, Ruedenberg K, Cheung LM (1972) J Chem Phys 57:2787
- 69. Implementing the SAAP formalism II. Simultaneous eigenfunctions of L^2 and S^2 by direct diagonalization. Safmon WI, Ruedenberg K (1972) J Chem Phys 57:2791
- Rotation and translation of regular and irregular solid spherical harmonics. Steinborn EO, Ruedenberg K (1973) in: Löwdin PO (ed) Quantum Chem 7:1
- Molecular integrals between real and between complex spherical harmonics. Steinborn EO, Ruedenberg K (1973) in: Löwdin PO (ed) Quantum Chem 7:83
- 72. Description of molecules in terms of localized orbitals. Ruedenberg K (1973) in: Herman F, McLean AD, Nesbet RK (eds) Computational Methods for Large Molecules and Localized States in Solids. Plenum Press, NY, p 149
- Nonorthogonal atomic self-consistent-field orbitals. Raffenetti RC, Ruedenberg K (1973) J Chem Phys 59:5950

- 74. Even-tempered atomic orbitals. III. Economic deployment of Gaussian primitives in expanding atomic SCF orbitals. Bardo RD, Ruedenberg K (1973) J Chem Phys 59:5956
- 75. Even-tempered atomic orbitals. IV. Atomic orbital bases with pseudo-scaling capability for molecular calculations. Bardo RD, Ruedenberg K (1973) J Chem Phys 59:5966
- 76. Even-tempered atomic orbitals. V. SCF calculations of trialkali ions with pseudo-scaled non-orthogonal bases. Raffenetti RC, Ruedenberg K (1973) J Chem Phys 59:5978
- 77. Why is the delocalization energy negative and why is it proportional to the number of double bonds? England W, Ruedenberg K (1973) J Am Chem Soc 95:8769
- Even-tempered atomic orbitals. VI. Optimal orbital exponents and optimal orbital contractions of Gaussian primitives for hydrogen, carbon and oxygen in molecules. Bardo RD, Ruedenberg K (1974) J Chem Phys 60:918
- Even-tempered atomic orbitals. VII. Theoretical equilibrium geometries and reaction energies for carbon suboxyde and other molecules containing carbon, oxygen and hydrogen. Bardo RD, Ruedenberg K (1974) J Chem Phys 60:932
- Transferable localized molecular orbitals for acyclic hydrocarbons. England W, Gordon MS, Ruedenberg K (1975) Theor Chim Acta 37:177
- 81. The nature of the chemical bond, an energetic view. Ruedenberg K (1975) in: Daudel R (ed) Localization and delocalization in quantum chemistry, Vol 1:222. Reidel, Dordrecht, Holland
- MCSCF studies of chemical reactions. I. Natural reaction orbitals and localized reaction orbitals. Ruedenberg K, Sundberg K (1976) in: Calais JL, Goscinski O, Linderberg J, Ohrn Y (eds) Quantum Science, Plenum, NY, p 505
- 83. Approximate relation between SCF orbital energies and total SCF energy. Ruedenberg K (1977) J Chem Phys 66:375
- 84. Molecular orbital bonding concepts in polyatomic molecules: A novel pictorial approach. Hoffman DK, Ruedenberg K, Verkade JG (1977) Structure and Bonding 33:57
- A novel pictorial approach to teach MO bonding concepts in polyatomic molecules. Hoffman DK, Ruedenberg K, Verkade JG (1977) J Chem Ed 54:590
- 86. Dimerization of carbene to ethylene. Cheung LM, Sundberg KR, Ruedenberg K (1978) J Am Chem Soc (Communication) 100:8024
- MCSCF optimization through combined use of natural orbitals and the Brillouin-Levy-Berthier theorem. Ruedenberg K, Cheung LM, Elbert ST (1979) Int J Quantum Chem 16:1069
- Electronic rearrangements during chemical reactions II. Planar dissociation of ethylene. Cheung LM, Sundberg KR, Ruedenberg K (1979) Int J Quantum Chem 16:1103
- 89. Systematic approach to extended even-tempered orbital bases for atomic and molecular calculations, Feller DF, Ruedenberg K (1979) Theor Chim Acta 52:231
- 90. Effective convergence to complete orbital bases and to the atomic Hartree-Fock limit through systematic sequences of Gaussian primitives. Schmidt MW, Ruedenberg K (1979) J Chem Phys 71:3951
- Are atoms intrinsic to molecular wave functions? I. The FORS model. Ruedenberg K, Schmidt MW, Gilbert MM, Elbert ST (1982) Chemical Physics 71:41
- 92. Are atoms intrinsic to molecular electronic wave functions? II. Analysis of FORS orbitals. Ruedenberg K, Schmidt MW, Gilbert M (1982) Chemical Physics 71:51
- 93. Are atoms intrinsic to molecular electronic wave functions? III. Analysis of FORS configurations. Ruedenberg K, Schmidt MW, Gilbert M, Elbert ST (1982) Chemical Physics 71:65
- 94. Concerted dihydrogen exchange between ethane and ethylene. SCF and FORS calculations of the barrier. Feller DF, Schmidt MW, Ruedenberg K (1982) J Am Chem Soc 104:960
- 95. Intra-atomic correlation correction in the FORS model. Lam MTB, Schmidt MW, Ruedenberg K (1985) J Phys Chem 89:2221
- 96. Chemical binding and electron correlation in diatomic molecules as described by the FORS model and the FORS-IACC model. Schmidt MW, Lam MTB, Elbert ST, Ruedenberg K (1985) Theor Chim Acta 68:69
- Representation of three-dimensional electron distributions through a perspective view of contour diagrams in a set of parallel planes. Gilbert MM, Donn JJ, Peirce M, Sundberg KR, Ruedenberg K (1985) J Comput Chem 6:209
- Electron difference densities and chemical binding. Valtazanos P, Schwarz WHE, Ruedenberg K (1985) Theor Chim Acta 68:471

- 99. Gradient extremals. Hoffman DK, Nord RS, Ruedenberg K (1986) Theor Chim Acta 69:265
- 100. Bifurcations and transition states. Ruedenberg K, Valtazanos P (1986) Theor Chim Acta 69:281
- 101. Ring opening of cyclopropylidenes to allenes: Reactions with bifurcating transition regions, free internal motions, steric hindrances and long-range dipolar interactions. Valtazanos P, Elbert ST, Ruedenberg K (1986) J Am Chem Soc 108:3147
- 102. Generation of a full active configuration space in terms of symmetry- and spin-adapted antisymmetrized orbital products. Lam MTB, Elbert ST, Ruedenberg K (1987) Int J Quantum Chem 31:489
- 103. X-ray diffraction, electron densities and chemical bonding. Schwarz WHE, Mensching L, Ruedenberg K, Jacobson R, Miller LL (1988) Portgal Phys 19:185
- Chemical deformation densities. I. Formulation and quantitative determination. Schwarz WHE, Ruedenberg K, Mensching L (1989) J Am Chem Soc 111:6926
- 105. Chemical deformation densities. II. Small molecules. Mensching L, von Niessen W, Valtazanos P, Ruedenberg K, Schwarz WHE (1989) J Am Chem Soc 111:6933
- 106. Electron densities, deformation densities, and chemical bonding, Schwarz WHE, Mensching L, Ruedenberg K, Miller LL, Valtazanos P, von Niessen W, Jacobson RA (1989) Angew Chem 101:605 and Int Ed Eng 28:597
- Nonspherical atom densities and chemical deformation densities from X-ray scattering. Ruedenberg K, Schwarz WHE (1990) J Chem Phys 92:4956
- 108. The potential energy surface of the ground state of carbon dioxide. Xantheas SS, Elbert ST, Ruedenberg K (1990) Chem Phys Lett 166:39
- 109. An intersection seam between the ground state of ozone and an excited state of like symmetry. Ruedenberg K, Xantheas S, Elbert ST (1990) J Chem Phys 93:7519
- 110. The ring opening of cyclopropylidene to allene: Global features of the reaction surface. Valtazanos P, Elbert ST, Xantheas S, Ruedenberg K (1991) Theor Chim Acta 78:287
- 111. The ring opening of cyclopropylidene to allene and the isomerization of allene: Ab-initio interpretation of the electronic rearrangements in terms of quasiatomic orbitals. Xantheas S, Valtazanos P, Ruedenberg K (1991) Theor Chim Acta 78:327
- 112. The ring opening of cyclopropylidene to allene: Key features of the accurate reaction surface. Valtazanos P, Xantheas S, Elbert ST, Ruedenberg K (1991) Theor Chim Acta 78:365
- 113. The ring opening of substituted cyclopropylidene to substituted allene: The effect of steric and longrange electrostatic interactions. Valtazanos P, Ruedenberg K (1991) Theor Chim Acta 78:397
- 114. Potential energy surfaces of ozone. Xantheas IS, Atchity G, Elbert ST, Ruedenberg K J Chem Phys, accepted
- 115. The intersection of two potential energy surfaces. Atchity G, Xantheas S, Ruedenberg K J Chem Phys, accepted
- 116. The ground state potential energy surface of carbon dioxide. Xantheas S, Elbert ST, Ruedenberg K J Chem Phys
- 117. Ring opening of the lowest triplet state of cyclopropylidene to allene. IV. Xantheas S, Elbert ST, Ruedenberg K Theor Chim Acta

Special reports and works under direction or supervision of Klaus Ruedenberg

- A1. Spectral characteristics of several series of more unusual aromatic hydrocarbons. Miller Layton E Jr (1960) J Mol Spectroscopy 5:181
- A2. Calculation on electronic spectra of catacondensed and pericondensed aromatic hydrocarbons. Hummel RL, Ruedenberg K (1964) USAEC Research And Development Report No. IS-450. Office of Technical Services, U.S. Dept. of Commerce, Washington DC
- A3. Deviations from the virial relationship in many-centered variational functions. Wasserman AL (1964) J Chem Phys 40:1812
- A4. Free-electron theory of conjugated molecules. Platt JR, Ruedenberg K, Scherr CW, Ham NS, Labhart H, Lichten W (1964) Wiley, NY (contains items No. 5, 6, 8, 9, 10, 11, 12, 15, 28 of this publication list)

- A5. The physical nature of the chemical bond. Ruedenberg K (1964) MIR Publ, Moscow (unauthorized paperback edition of unauthorized Russian translation of item No. 19 of this publication list)
- A6. Robert S Mulliken, Nobelpreisträger für Chemie 1966. Ruedenberg K (1966) Neue Züricher Zeitung 187(4937):1
- A7. Two-centered Coulomb and hybrid integrals. Miller KJ (1968) J Math Phys 9:1292
- A8. Natural orbital expansion of interacting geminals. Silver DM (1969) J Chem Phys 50:5108
- A9. Energy localization of approximate localized orbitals. England W, Gordon MS (1969) J Am Chem Soc 91:6864
- A10. Poly-polar expansions for regular and irregular spherical harmonics in molecules. Steinborn O (1969) Chem Phys Lett 3:671
- A11. Influence of long-range Coulombic interactions on binding energy curves of molecular ions. Feinberg MJ (1970) Theoret Chim Acta 19:109
- A12. Continuous degeneracy and energy localization of molecular orbitals. England W (1971) Int J Quantum Chem 5:683
- A13. Localized charge distributions. I. General theory, energy partitioning and the internal rotation barrier in ethane. England W, Gordon MS (1971) J Am Chem Soc 93:4649
- A14. One-center Coulomb, two-center hybrid and two-center Coulomb integrals for Slater-transform-Preuss atomic orbitals. England W (1972) Int J Quantum Chem 6:509
- A15. Localized charge distribution. II. An interpretation of the barriers to internal rotation in H₂O₂. England W, Gordon MS (1972) J Am Chem Soc 94:4818
- A16. Localized charge distribution. III. Transferability and trends of CH moments and energies in acyclic hydrocarbons. Gordon MS, England W (1972) J Am Chem Soc 94:5168
- A17. Localized charge distribution. IV. The integral rotation barrier in borazene. Gordon MS, England W (1972) Chem Phys Lett 15:59
- A18. Even-tempered atomic orbitals. II. Atomic self-consistent-field wavefunctions in terms of even-tempered exponential bases. Raffenetti RC (1973) J Chem Phys 59:5936
- A19. Convergence of the completely separated bipolar expansion of r_{12}^{-1} . Salmon LS (1973) Int J Quantum Chem 7:411
- A20. Localized charge distribution. V. The internal rotation barriers in methylamine, methyl alcohol, propene, and acetaldehyde. Gordon MS, England (1973) J Am Chem Soc 95:1753
- A21. First-order scheme for energy localization. England W (1973) J Chem Phys 58:5182
- A22. Even-tempered orbital bases for atoms and molecules. Ruedenberg K, Raffenetti RC, Bardo RD (1973) in: Smith DW (ed) Proc 1972 Boulder Summer Research Conf on Theoretical Chemistry. Wiley, NY, p 164
- A23. Even-tempered exponential representation of atomic self-consistent-field wavefunctions. Raffenetti RC, Ruedenberg K (1973) USAEC Research and Development Report No. IS 3195, Ames Laboratory, USAEC, Iowa State University
- A24. Genealogical spin eigenfunctions and antisymmetric many-electron wavefunctions generated directly from Young diagrams. Salmon WI (1974) in: Löwdin PO (ed) Adv Quantum Chem 8:37, Academic Press, NY
- A25. Pictorial representation of three-dimensional distributions through a perspective view of contour diagrams in a set of parallel planes. Dombek MG, Donn JJ, Sundberg KR, Ruedenberg K (1974) USAEC Research and Development Report No. IS 3329, Ames Laboratory, USAEC, Iowa State University
- A26. Potential energy curve of the *cis-trans*-isomerization of glyoxal. Sundberg KR, Cheung LM (1974) Chem Phys Lett 29:93
- A27. Determination of orbitals and selection of configurations through the method of the full optimized reaction space. Ruedenberg K (1979) in: Report on the NRCC 1978 Workshop on Post-Hartree-Fock Quantum Chemistry, p. 46. Lawrence Berkeley Laboratory, Univ of California, Report LBL 8233, UC4, Conf 780883
- A28. MCSCF optimization through iterative CI calculations in the single excitation space and MCSCF wavefunctions in the full reaction space. Ruedenberg K (1981) in Proc NRCC Workshop on Recent Developments of Multiconfiguration Hartree-Fock Methods, July 15-17, 1981 at Texas A&M University. Lawrence Berkeley Laboratory, Univ of California, p 51

- A29. The sudden polarization effect: MCSCF calculations on planar and 90° twisted methylenecyclopropane. Johnson RP, Schmidt MW (1981) J Am Chem Soc 103:3244
- A30. Small ring cyclic allenes: An *ab-initio* study of the structure of 1,2-cyclohexadiene. Schmidt MW, Angus RO, Johnson RP (1982) J Am Chem Soc 104:6838
- A31. Polarized nonvertical excited states: FORS MCSCF and CI study of torsion and bending in allene. Lam B, Johnson RP (1983) J Am Chem Soc 105:7479
- A32. Small ring cyclic cumulenes: Theoretical studies of the structure and barrier to inversion in cyclic allenes. Angus RO, Schmidt MW, Johnson RP (1985) J Am Chem Soc 107:532
- A33. A chemically useful definition of electron difference densities. Schwarz WHE, Mensching L, Valtazanos P, von Niessen W (1986) Int J Quantum Chem 29:909
- A34. International evaluation of theoretical chemistry in Sweden. Buckingham AD, Manne R, Rowlinson JS, Ruedenberg K, Warshel A, Norden B (1988) Swedish Natural Science Research Council